

## **Natural Products**

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## Total Synthesis of Schilancitrilactones B and C\*\*

2: Schilancitrilactones C

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Abstract: The first total syntheses of schilancitrilactones B and C have been accomplished in 17 steps (longest linear sequence) from commercially available materials. Key steps include an intramolecular radical cyclization to provide the seven-membered ring, late-stage iodination, and an intermolecular radical addition reaction to complete the total synthesis.

**S**chilancitrilactones B and C (**1** and **2**; Figure 1)<sup>[1]</sup> were isolated in 2012 by Sun and co-workers from the stems of *Schisandra Lancifolia*, which have been used in traditional Chinese medicine for the treatment of neurasthenia and related diseases.<sup>[2]</sup> Preliminary biological assays indicated that

O H HO H A B C D A B H HO D O A

1: Schilancitrilactones B

Figure 1. Schilancitrilactones B and C.

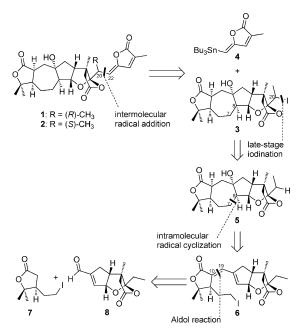
schilancitrilactone C showed biological activities for inhibiting HIV-1 while schilancitrilactone B was not bioactive. The structures of these compounds were striking in that they contain a 5/7/5/5-fused pentacyclic ring system bearing nine stereogenic centers. In addition, the three *cis*-fused five-membered rings (rings C–E), all with the envelope conformations, and seven contiguous chiral centers (including two quaternary centers) form a structurally rigid tricyclic ring system. Construction of these highly oxygenated unique motifs remains challenging. Herein, we present the first total synthesis of schilancitrilactones B and C. The key steps include the successful implementation of an intramolecular radical cyclization to prepare a seven-member ring, late-stage iodination, and an intermolecular radical C–C bond formation.

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Recently, the total synthesis of *Schisandraceae* triterpenoids has been of great interest to synthetic organic chemists because of the intriguing structures and diverse biological activities.<sup>[3]</sup> In 2011, Yang and co-workers reported the first total synthesis of schindilactone A.<sup>[4]</sup> Recently, the group of Li disclosed the first asymmetric total synthesis of rubriflordilactone A.<sup>[5]</sup> Herein we report our efforts on developing a new strategy to solve the chemical synthesis of 1 and 2, and a pathway for the synthesis of their analogues and derivatives for medicinal studies. Our retrosynthetic analysis is shown in Scheme 1. It was hypothesized that 1 and 2 might be



Scheme 1. Retrosynthetic analysis of schilancitrilactones B and C.

accessible by an intermolecular radical addition reaction between the alkyl iodide 3 and vinyl stannane 4. The alkyl iodide 3 was expected to arise by late-stage iodination at C20 from the compound 5, which in turn could be prepared from the compound 6 by a series of steps including an intramolecular radical cyclization at the C7–C8 bond to prepare the seven-membered ring. The compound 6 was further deconstructed at the C10–C19 bond into the two simple building blocks 7 and 8, which could be put together by an aldol reaction. The building blocks 4, 7, and 8 could be prepared from the commercially available compounds citraconic anhydride (16), L-carvone (9), and 1,3-cyclohexadiene (19), respectively.

Our work began with the synthesis of the alkyl iodide **7** (Scheme 2). Following the procedure by Fukuyama and coworkers.<sup>[7]</sup> L-carvone (**9**) was converted into the correspond-



Scheme 2. Reagents and conditions: a) 30% H<sub>2</sub>O<sub>2</sub>, NaOH (aq), MeOH, 0°C; b) H<sub>2</sub>SO<sub>4</sub>, THF/H<sub>2</sub>O (5:1), reflux; c) NaIO<sub>4</sub>, iPrOH/H<sub>2</sub>O (1:1), 0°C to RT; d) I<sub>2</sub>, KI, NaHCO<sub>3</sub>, CH<sub>2</sub>CI<sub>2</sub>/H<sub>2</sub>O (1:3), 0°C, 51% for 4 steps; e) NaBH<sub>4</sub>, MeOH, 0°C, 85%; f) AIBN, Bu<sub>3</sub>SnH, toluene, 100°C, 90%; g) PPh<sub>3</sub>, I<sub>2</sub>, imidazole, 0°C to RT, THF, 84%. AIBN = 2,2'-azobis (2-methylpropionitrile), THF = tetrahydrofuran.

ing aldehyde 13 in a four-step sequence involving epoxidation, epoxide hydrolysis, oxidative cleavage of diols, and iodolactonization in 51 % overall yield (4 steps). The aldehyde 13 was selectively reduced with NaBH<sub>4</sub> to provide the alcohol 14 in 85% yield. The deiodination of 14 with AIBN and Bu<sub>3</sub>SnH afforded the compound 15, which was converted into the corresponding 7 with I<sub>2</sub> in the presence of Ph<sub>3</sub>P and imidazole in 84% yield.[8]

Depicted in Scheme 3 is the construction of the vinyl stannane compound 4. The vinyl bromide 18 was prepared

Scheme 3. Reagents and conditions: a) PPh<sub>3</sub>CHCO<sub>2</sub>tBu, toluene, RT, 54%; b) TFA,  $CH_2Cl_2$ , 0°C; c)  $Br_2$ , TFA,  $CDCl_3/CCl_4$  (1:1), RT; (d)  $Et_3N$ , DMF, 0°C to RT, 76% for 3 steps; (e) [{Pd(allyl)Cl}<sub>2</sub>] (5 mol%), (Bu<sub>3</sub>Sn)<sub>2</sub>, LiCl, 1,4-dioxane, RT, 49%. DMF = N,N-dimethylformamide, TFA = trifluoromethanesulfonyl.

from the commercially available compound citraconic anhydride (16) in a reported four-step process in a 41% overall yield.<sup>[9]</sup> Stannylation of 18 was achieved and afforded 4 with [{Pd(allyl)Cl}<sub>2</sub>] and (Bu<sub>3</sub>Sn)<sub>2</sub> in 49 % yield.<sup>[10]</sup> It is noteworthy that 4 is not stable during purification, thus resulting in a low

We then moved on to construct the aldehyde compound 8 (Scheme 4). By using the reaction conditions developed by

Scheme 4. Reagents and conditions: a) TPP, O2, hv, CCl4, -10°C then thiourea, MeOH; b) BzCl, Et<sub>2</sub>N, DMAP, CH<sub>2</sub>Cl<sub>2</sub>, RT, 73% for 2 steps; c)  $[\{Pd(allyl)Cl\}_2]$  (3 mol%), ligand A, 22,  $K_2CO_3$ , MeOH, THF, 0°C; then DIPEA, 55°C; 70%; d) NaH, CH<sub>3</sub>I, DMF, 0°C, 90%, d.r. (at C13) = 4:1; e) NaBr, DMF, 180°C, 88%, d.r. (at C13) = 1:1; f) LDA,  $BrCH_2CO_2tBu$ , THF, -78 °C; g) TFA,  $CH_2CI_2$ , 0 °C to RT, 93 % for 2 steps; h) EtMgBr, THF/Et<sub>2</sub>O (1:1), -78 °C to RT, 80%; i) O<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>/ MeOH (1:1), -78 °C, then Me<sub>2</sub>S, -78 °C to RT; j) (Bn<sub>2</sub>NH<sub>2</sub>) (OCOCF<sub>3</sub>), toluene, 63 °C, 80% for 2 steps. Bz = benzoyl, DIPEA = diisopropylethylamine, DMAP = 4-(N, N-dimethylamino) pyridine, LDA = lithiumdiisopropyl amide, TPP=5,10,15,20-tetraphenyl-21H,23H-porphine. Thermal ellipsoids are shown at 50% probability.[21]

Trost and co-workers.[11] the lactone 23 was obtained in a reported three-step process from the commercially available 1,3-cyclohexadiene (19). The steps included asymmetric palladium-catalyzed allylic alkylation. Methylation of 23 with NaH and CH<sub>3</sub>I provided the compound 24 in 90% yield with 4:1 diastereoselectivity at C13, and was then subjected to decarboxylation mediated by NaBr to produce a 1:1 mixture of the lactone 25 in 88 % yield. Alkylation of 25 with *tert*-butyl bromoacetate gave the single diastereomer **26**. Deprotection of 26 was achieved using trifluoroacetic acid and gave the acid 27 in 93% yield (two steps). Addition of ethyl magnesium bromide followed by acidic workup gave rise to the tricycle 28, having an ethyl group installed stereoselectively onto the tricyclic framework. [12] The absolute configuration of 28 was determined by X-ray crystallographic analysis. The cyclohexene ring in 28 was oxidatively cleaved by ozonolysis and the resulting dialdehyde 29 was directly subjected to intramolecular aldol condensation, thus yielding the ring-closed unsaturated aldehyde 8 (80% yield for two steps).[13]

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**Scheme 5.** Reagents and conditions: a) LDA, THF, -78 °C, then **8**, 86%, d.r. (at C19) = 17:1; b) CuCl<sub>2</sub>, EDC, toluene, 80 °C, 83 %; c) CuI, Zn, Pyr/H<sub>2</sub>O (1:4), ultrasound, RT, 55% for **31**, 4% for **31**′; d) mCPBA, NaHCO<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, -15 °C, 51%; e) NaOMe, MeOH, RT; then NiCl<sub>2</sub>·6 H<sub>2</sub>O, NaBH<sub>4</sub>, MeOH/THF (1:5), -15 °C, 73%; (f) ICl, THF, RT, 63%, d.r. (at C20) = 1.5:1; (g) **4**, AIBN, Bu<sub>3</sub>SnH, toluene, 4 Å M.S., 100 °C, 9% for **1**, 36% for **2**. EDC = 1-(3-dimethylaminopropyl)-3-ethylcarbodiimidem mCPBA = m-chloroperbenzoic acid. Thermal ellipsoids are shown at 50% probability. [21]

With the key intermediates 4, 7, and 8 in hand, we finished the total synthesis of schilancitrilactones B and C as shown in Scheme 5. The iodo compound 7 was converted into the lithium enolate with LDA at -78 °C and then reacted with 8 to give the aldol adduct 30 in 86% yield (d.r. = 17:1 at C19). Dehydration of 30 with 2 equivalents of EDC and a catalytic amount of CuCl<sub>2</sub> provided a 2:1 mixture of the inseparable diene lactone 6 in 83 % yield. [14] The structure of the E isomer was confirmed by X-ray crystallographic analysis. Then we investigated the intramolecular radical cyclization to form the seven-membered ring. Initially, the conventional radical conditions (AIBN, Bu<sub>3</sub>SnH) led to rapid decomposition of 6 and trace amounts of cyclization product was observed. Photoredox catalysis<sup>[15]</sup> was also evaluated and no desired product was found. By using the method (CuI, Zn under ultrasound) for conjugate additions in aqueous media discovered by Luche et al., [16] we were pleased to observe the cyclization product 31 in 55 % yield, together with the isomer 31' in 4% yield. The structure of 31' was confirmed by X-ray crystallographic analysis. We reasoned that the conformation of 6 was suited for cyclization to give the seven-membered ring over the five-membered ring.<sup>[17]</sup> Epoxidation of **31** with mCPBA provided the epoxide 32 in 51% yield, and underwent ring opening with NaOMe/NiCl<sub>2</sub>·6H<sub>2</sub>O/NaBH<sub>4</sub> to give the alcohol 5 in 73%.[18] During this transformation, the epoxide 32 was converted into the intermediate 33 with NaOMe and then further reduced to give the desired product 5 with NiCl<sub>2</sub>·6H<sub>2</sub>O and NaBH<sub>4</sub>. Finally, we investigated the late-stage iodination and intermolecular radical addition reaction. It was found that treatment of 5 with ICl delivered the iodo compound **3** as a mixture of diastereomers (d.r. = 1.5:1 at C20) in 63 % yield, [18] and when **3** was heated with the vinyl stannane **4**, AIBN, and Bu<sub>3</sub>SnH provided the schilancitrilactones B (**1**, 9%) and C (**2**, 36%) in 45% total yield. Around 25% yield of other isomers were observed based on the HNMR analysis of the crude reaction mixture. [20] The characterization data obtained for synthetic **1** and **2** were in accord with the reported data for the natural products.

In summary, the first total synthesis of schilancitrilactones B and C has been accomplished by employing an intramolecular radical cyclization, late-stage iodination, and intermolecular radical addition as key steps in the 17 step synthesis (longest linear sequence) from commercially available materials. This strategy opens a pathway for the syntheses of other compounds related to schilancitrilactones B and C, as well as their derivatives and analogues.

**Keywords:** cyclizations  $\cdot$  natural products  $\cdot$  radical chemistry  $\cdot$  total synthesis  $\cdot$  terpenoids

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<sup>[1]</sup> X. Luo, Y.-M. Shi, R.-H. Luo, S.-H. Luo, X.-N. Li, R.-R. Wang, S.-H. Li, Y.-T. Zheng, X. Du, W.-L. Xiao, J.-X. Pu, H.-D. Sun, Org. Lett. 2012, 14, 1286.

<sup>[2]</sup> a) W.-L. Xiao, R.-T. Li, S.-X. Huang, J.-X. Pu, H.-D. Sun, *Nat. Prod. Rep.* 2008, 25, 871; b) Y.-M. Shi, W.-L. Xiao, J.-X. Pu, H.-D. Sun, *Nat. Prod. Rep.* 2015, 32, 367.

<sup>[3]</sup> Selected examples of synthetic efforts on Schisandraceae triterpenoids: a) Y. Tang, L. Deng, Y. Zhang, G. Dong, J.



- Chen, Z. Yang, Org. Lett. 2005, 7, 593; b) Y. Tang, Y. Zhang, M. Dai, T. Luo, L. Deng, J. Chen, Z. Yang, Org. Lett. 2005, 7, 885; c) Y. Zhang, Y. Tang, T. Luo, J. Shen, J. Chen, Z. Yang, Org. Lett. **2006**, 8, 107; d) D. Fischer, E. A. Theodorakis, Eur. J. Org. Chem. 2007, 4193; e) Y. Zhang, W. Ren, Y. Lan, Q. Xiao, K. Wang, J. Xu, J. Chen, Z. Yang, Org. Lett. 2008, 10, 665; f) Q. Wang, C. Chen, Org. Lett. 2008, 10, 1223; g) L. A. Paquette, K. W. Lai, Org. Lett. 2008, 10, 2111; h) K. W. Lai, L. A. Paquette, Org. Lett. 2008, 10, 2115; i) L. A. Paquette, K. W. Lai, Org. Lett. 2008, 10, 3781; j) M. F. Hossain, K. Matcha, S. Ghosh, Tetrahedron Lett. 2011, 52, 6473; k) A. Bartoli, G. Chouraqui, J.-L. Parrain, Org. Lett. 2012, 14, 122; l) V. A. Ignatenko, Y. Han, G. P. Tochtrop, J. Org. Chem. 2013, 78, 12229; m) B. Gockel, S. S. Goh, E. J. Puttock, H. Baars, G. Chaubet, E. A. Anderson, Org. Lett. 2014, 16, 4480.
- [4] a) Q. Xiao, W.-W. Ren, Z.-X. Chen, T.-W. Sun, Y. Li, Q.-D. Ye, J.-X. Gong, F.-K. Meng, L. You, Y.-F. Liu, M.-Z. Zhao, L.-M. Xu, Z.-H. Shan, Y. Shi, Y.-F. Tang, J.-H. Chen, Z. Yang, Angew. Chem. Int. Ed. 2011, 50, 7373; Angew. Chem. 2011, 123, 7511; b) T.-W. Sun, W.-W. Ren, Q. Xiao, Y.-F. Tang, Y.-D. Zhang, Y. Li, F.-K. Meng, Y.-F. Liu, M.-Z. Zhao, L.-M. Xu, J.-H. Chen, Z. Yang, Chem. Asian J. 2012, 7, 2321; c) Y. Li, Z.-X. Chen, Q. Xiao, Q.-D. Ye, T.-W. Sun, F.-K. Meng, W.-W. Ren, L. You, L.-M. Xu, Y.-F. Wang, J.-H. Chen, Z. Yang, Chem. Asian J. 2012, 7, 2334; d) W.-W. Ren, Z.-X. Chen, D. Xiao, Y. Li, T.-W. Sun, Z.-Y. Zhang, Q.-D. Ye, F.-K. Meng, L. You, M.-Z. Zhao, L.-M. Xu, Y.-F. Tang, J.-H. Chen, Z. Yang, Chem. Asian J. 2012, 7, 2341.
- [5] J. Li, P. Yang, M. Yao, J. Deng, A. Li, J. Am. Chem. Soc. 2014, 136, 16477.
- [6] Selected reviews for radical cyclization: a) J. M. Cuerva, J. Justicia, J. L. Oller-López, J. E. Oltra, Top. Curr. Chem. 2006, 264, 63; b) J. Justicia, L. Cienfuegos, A. G. Campaña, D. Miguel, V. Jakoby, A. Gansäuer, J. M. Cuerva, Chem. Soc. Rev. 2011, 40, 3525; c) T. Taniguchi, H. Ishibashi, Heterocycles 2013, 87, 527. For selected example, see: J. Justicia, J. L. Oller-López, A. G. Campaña, J. E. Oltra, J. M. Cuerva, E. Buñuel, D. J. Cárdenas, J. Am. Chem. Soc. 2005, 127, 14911.
- [7] a) S. Takita, S. Yokoshima, T. Fukuyama, Org. Lett. 2011, 13, 2068; b) S. Takita, S. Yokoshima, T. Fukuyama, Synthesis 2011, 23, 3848.
- [8] E. J. Corey, S. G. Pyne, W.-G. Su, Tetrahedron Lett. 1983, 24,

- [9] T. Benneche, Z. Hussain, A. A. Scheie, J. Lönn-Stensrud, New J. Chem. 2008, 32, 1567.
- [10] W. D. Wulff, G. A. Peterson, W. E. Bauta, K.-S. Chan, K.-L. Faron, S. R. Gilbertson, R. W. Kaesler, D. C. Yang, C. K. Murray, J. Org. Chem. 1986, 51, 277.
- [11] a) B. M. Trost, S. Tanimori, P. T. Dunn, J. Am. Chem. Soc. 1997, 119, 2735; b) Y.-H. Wang, L.-L. Zhu, Y.-X. Zhang, Z. Chen, Chem. Commun. 2010, 46, 577.
- [12] K. Harrar, O. Reiser, Chem. Commun. 2012, 48, 3457.
- [13] a) E. J. Corey, R. L. Danheiser, S. Chandrasekaran, P. Siret, G. E. Keck, J.-L. Gras, J. Am. Chem. Soc. 1978, 100, 8031; b) G. Coulthard, W. Erb, V. K. Aggarwal, Nature 2012, 489, 278.
- [14] a) H. Sai, H. Ohmizu, Tetrahedron Lett. 1999, 40, 5019; b) H. Sai, T. Ogiku, H. Ohmizu, Tetrahedron 2007, 63, 10345.
- [15] For selected reviews, see: a) K. Zeitler, Angew. Chem. Int. Ed. 2009, 48, 9785; Angew. Chem. 2009, 121, 9969; b) T. P. Yoon, M. A. Ischay, J. Du, Nat. Chem. 2010, 2, 527; c) J. M. R. Narayanam, C. R. J. Stephenson, Chem. Soc. Rev. 2011, 40, 102; d) J. Xuan, W. Xiao, Angew. Chem. Int. Ed. 2012, 51, 6828; Angew. Chem. 2012, 124, 6934; e) C. K. Prier, D. A. Rankic, D. W. C. MacMillan, Chem. Rev. 2013, 113, 5322.
- [16] J. L. Luche, C. Allavena, Tetrahedron Lett. 1988, 29, 5369.
- [17] a) E. Lee, C. H. Yoon, T. H. Lee, J. Am. Chem. Soc. 1992, 114, 10981; b) E. Lee, C. H. Yoon, Tetrahedron Lett. 1996, 37, 5929; c) C. S. Swindell, M. C. Chander, J. M. Heerding, P. G. Klimko, L. T. Rahman, J. V. Raman, H. Venkataraman, J. Org. Chem. **1996**, 61, 1101.
- [18] T. Satoh, K. Nanba, S. Suzuki, Chem. Pharm. Bull. 1971, 19, 817.
- [19] a) G. Giordano, L. Coppi, A. Restelli, J. Org. Chem. 1990, 55, 5400; b) P. S. Baran, A. L. Zografos, D. P. O'Malley, J. Am. Chem. Soc. 2004, 126, 3726.
- [20] J. E. Baldwin, D. R. Kelly, J. Chem. Soc. Chem. Commun. 1985, 682.
- [21] CCDC 1034710 (28), 1046754 (6), and 1034713 (31') contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data\_request/cif.

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